	Potassium sorbate	March 2004 Page 1 of 2
Section A4.1 Annex Point IIA4.1 IIIA- IV.1	Analytical Methods for Detection and Identification	
	1 REFERENCE	Official use only
Reference	A4.1/01: Verfahren zur Prüfung der Gehaltsbestimmung von Sorbinsäure, Kaliumsorbat, Kaliumsorbat Lösung 50 % und Calciumsorbat, A4.1/02: Verfahren zur Bestimmung des Wassergehaltes in Kaliumsorbat Pulver und Granulat sowie Calciumsorbat und Sorbinsäure,	
	The details of the analytical method are considered to represent commercially sensitive and therefore confidential data by	
	The report is summarised in detail in the file of confidential information of <b>Example</b> .	n
	Evaluation by Competent Authorities	1



		Potassium Sorbate	March 2004
			Page 1 of 10
Section A4.2 A Annex Point IIA4.2 tl		Analytical methods for detection and identification of the active substance in (a) soil	
			2040 000
		1 REFERENCE	Official use only
1.1	Reference	A4.2/01:	
		determination of Sorbic acid in soil.	•
1.2	Data protection	Yes	
1.2.1	Data owner		
1.2.2	Companies with letter of access	No	
1.2.3	Criteria for data protection	Data submitted to the MS after 13 May 2000 on existing a.s. for the purpose of its entry into Annex I/IA.	
		2 GUIDELINES AND QUALITY ASSURANCE	
2.1	Guideline study	SANCO/825/00 rev. 6.	
2.2	GLP	Yes (certified laboratory)	
2.3	Deviations	No	
		3 MATERIALS AND METHODS	3
3.1	Preliminary treatment		
3.1.1	Enrichment	Residues of Sorbic acid were extracted with acetonitrile/water (1/1), filtrated and evaporated to dryness.	
3.1.2	Clean-up	Further clean-up was performed by water steam distillation using an automated water steam distillation device and solid-phase extraction the acidified distillate on PPL-SPE-cartridges. Residues were eluted with methanol. After evaporation, residues were derivatised using $BF_3$ /methanol solution.	of
3.2	Detection		
3.2.1	Separation method	Capillary gas chromatography with an mass selective detector (GC-MSD).	
3.2.2	Detector	Mass selective detector ( $m/z = 111$ for quantification and $m/z = 126$ , for verification).	95
3.2.3	Standard(s)	Sorbic acid, batch no. 5661, purity 99.8%	
		Soil standard: Soil 3A (LUFA Speyer, Germany), batch no. F3A030 pH 7.1, dry matter 90%, organic carbon content 2.6%, soil type loan	3.

Section A4.2	Analytical methods for detection and identification of
Annex Point IIA4.2	the active substance in (a) soil

3.3	Linearity	
3.3.1	Calibration range	The detector response for the analytical standard was found to be linear between 0 and 1000 ng/ml in sample extracts.
3.3.2	Number of measurements	The calibration curve was plotted based on seven different concentrations.
3.3.3	Linearity	The equation of a typical standard calibration function for Sorbic acid was determined as (graph forced through the origin)
		$y = 0.128 x^2 + 354 x$ with a regression coefficient of $r^2 = 1.000$
		where $y$ is the response in the chromatogram, and $x$ the concentration of the substance [ng/ml].
3.4	Specificity: interfering substances	The method is suitable for the specific determination of residues of Sorbic acid in soil. As three fragment ions can be used for the determination, the method can be classified as highly specific. Under the chromatographic conditions used in this study, the retention time was approx. 5.6 min for Sorbic acid.
		Analysed blank control samples gave no interfering signals (< 30% of LOQ).
3.5	Recovery rates and standard deviations at different levels	The results are presented in Table A4.2-1.
3.6	Limit of determination	The limit of quantification (LOQ) is 0.05 mg/kg.
3.7	Precision	
3.7.1	Repeatability	The average recovery was in a range of 70 and 110%, the relative standard deviation was below 20%.
3.7.2	Independent laboratory validation	Not required.
		4 APPLICANT'S SUMMARY AND CONCLUSION
4.1	Materials and methods	Residues of Sorbic acid were extracted with acetonitrile/water (1/1). Further clean-up was performed by water steam distillation and solid- phase extraction on PPL-SPE-cartridges. After derivatisation with BF <sub>3</sub> in methanol, determination was performed by GC-MSD.
4.2	Conclusion	Average recoveries were in the range of 70 - 110% with relative standard deviations below 20%. No interfering blanks were observed. Therefore, this method fulfils the requirements of SANCO/825/00 rev.6 and can be used as an enforcement method for the determination of residues of Sorbic acid in soil.
		Since the analytically relevant moiety of Sorbic acid as well as of Potassium sorbate is the Sorbate anion, the submitted method is likewise recommended as a suitable technique for detection and identification of Potassium sorbate residues.

		Potassium Sorbate	March 2004
			Page 3 of 10
Section A4.2 Annex Point II	2 IA4.2	Analytical methods for detection and identification of the active substance in (a) soil	
4.2.1 Reliab	ility	1	
4.2.2 Defici	encies	None	ļ
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Endler the level level at			Recovery [%	6	
Fortification level [mg/kg]	n	Range	Mean ± RSD	Overall ± RSD	
0.05	5	84 - 123	99 ± 20%	96±16%	
0.5	5	77 - 108	97 ± 13%		

Table A4.2- 1: Recovery rates of Sorbic acid in soil.

RSD - relative standard deviation

n - number of determinations

	Potassium Sorbate	March 2004 Page 5 of 10
Section A4.2 Annex Point IIA4.2	Analytical methods for detection and identification of the active substance in (b) air	
	JUSTIFICATION FOR NON-SUBMISSION OF DATA	Official use only
Other existing data	Technically not feasible     Scientifically unjustified  X	
Limited exposure [X]	Other justification	
Detailed justification:	A method for the detection of residues in air is not suggested, since Potassium sorbate is neither volatile nor intended to be sprayed or applied in any other way resulting in occurrence of Potassium sorbate is air.	in

	Evaluation by Competent Authorities				
	Use separate "evaluation boxes" to provide transparency as to the comments and views submitted				
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		Potassium Sorbate	March 2004
		Fotassian Solbate	age 6 of 10
Section Annex	on A4.2 Point IIA4.2	Analytical methods for detection and identification of the active substance in (c) water	
		1 REFERENCE	Official use only
1.1	Reference	A4.2/02:	
		determination of Sorbic acid in surface water.	
1.3	Data protection	Yes	
1.2.1	Data owner		
1.2.3	Companies with letter of access	No	
1.2.3	Criteria for data protection	Data submitted to the MS after 13 May 2000 on existing a.s. for the purpose of its entry into Annex I/IA.	
		2 GUIDELINES AND QUALITY ASSURANCE	
2.1	Guideline study	SANCO/825/00 rev. 6.	
2.2	GLP	Yes (certified laboratory)	
2.3	Deviations	No	
		3 MATERIALS AND METHODS	
3.1	Preliminary (reatment		
3.1.1	Enrichment	Residues of Sorbic acid were extracted by solid-phase extraction of the acidified sample on PPL-SPE-cartridges. Residues were eluted with methanol and evaporated.	
3.1.2	Clean-up	No further clean up was performed. Residues of Sorbic acid were derivatised using BF <sub>3</sub> /methanol solution.	
3.2	Detection		
3.2.1	Separation method	Capillary gas chromatography with an mass selective detector (GC-MSD)	
3.2.2	Detector	Mass selective detector $(m/z = 111$ for quantification and $m/z = 126, 9$ for verification).	5

		Potassi	um Sorbate	March 2004 Page 7 of 10
Section A4.2 Annex Point IIA4.2		Analytical methods the active substanc	s for detection and identification o e in (c) water	ſ
3.2.3	Standard(s)	Sorbic acid, batch no. 50	661, purity 99.8%.	
		Specification of surface 'Federbach' in Malsch/0	water (sample was taken from the stream Germany):	
		Appearance:	slightly brown and weakly turbid, odourl	ess
		pH (20°C):	7.5	
		Specific electric conductivity (20°C):	120 µS/cm	
		Total hardness:	0.42 mmol/l	
		Spectral absorption coefficient (254 nm):	$12.6 \text{ m}^{-1}$	
		Dissolved organic carbo	on: 4.4 mg C/l.	
3.3	Linearity			
3.3.1	Calibration range	The detector response for between 0 and 250 ng/m	or the analytical standards was found to be l hl.	inear
3.3.4	Number of measurements	The calibration curve was based on seven different concentrations.		
3.3.5	Linearity	The equation of a typica was determined as (grap	The equation of a typical standard calibration function for Sorbic acid was determined as (graph forced through the origin)	
		y = 2090 x with a regres	sion coefficient of $r^2=0.999$	
		where $y$ is the response the substance [ng/ml].	in the chromatogram, and x the concentration	on of
3.4	Specificity: interfering substances	The method is suitable f sorbic acid in water. As determination, the meth- chromatographic condit approx. 5.6 min for sorb interfering signals (< 30	for the specific determination of residues of three fragment ions could be used for od can be classified as highly specific. Under ions used in this study, the retention time we bic acid. Blank control samples analysed gav % of LOQ).	er the as ve no
3.5	Recovery rates and standard deviations at different levels	Please refer to Table A4	1.2- 2.	
3.6	Limit of determination	The limit of quantificati	on (LOQ) is 0.1 µg/l.	
3.7	Precision			
3.7.1	Repeatability	The average recovery is standard deviation is les	in a range between $70 - 110\%$ and the relates than 20%.	tive
3.7.2	Independent laboratory validation	Not required.		
		4 APPLICANT	"S SUMMARY AND CONCLUSIC	N

		Potassium Sorbate	March 2004 Page 8 of 10
Section Annex	on A4.2 Point IIA4.2	Analytical methods for detection and identification of the active substance in (c) water	
4.1	Materials and methods	Residues of sorbic acid were extracted by solid-phase extraction on PPL-SPE-cartridges. Residues were eluted with methanol. After derivatisation with BF <sub>3</sub> in methanol determination was performed by GC-MSD.	
4.2	Conclusion	Average recoveries were in the range of $70 - 110\%$ with relative standard deviations < 20%. No interfering blanks were observed.	
		Therefore, this method fulfils the requirements of SANCO/825/00 rev. and can be used as an enforcement method for the determination of residues of sorbic acid in surface, ground and drinking water.	6
		Since the analytically relevant moiety of Sorbic acid as well as of Potassium sorbate is the Sorbate anion, the submitted method is likewi recommended as a suitable technique for detection and identification o Potassium sorbate residues.	se f
4.2.1	Reliability	1	
4.2.2	Deficiencies	None	

	Potassium Sorbate	March 2004
		Page 9 of 10
	Evaluation by Competent Authorities	<i>K</i> - 1
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Table A4.2- 2: Recovery rate	s of Sorbic acid in surface water.
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Fortification level  µg/l		Recovery  %]			
	n	Range	Mean ± RSD	Overall ± RSD	
0.1*	5	86 - 122	100 ± 14%	90 ± 17%	
1.0	5	67 – 91	81 ± 12%		

RSD: relative standard deviation n: number of determinations

	Potassium Sorbate Ma	e 10 of 1
Section A4.2 Annex Point IIA4.2	Analytical methods for detection and identification of the active substance in (d) animal and human body fluids and tissues	
	JUSTIFICATION FOR NON-SUBMISSION OF DATA	Offici use on
Other existing data	Technically not feasible     Scientifically unjustified  X	
Limited exposure  X	Other justification [ ]	
Detailed justification:	A method for the detection of residues of t he active substance in animal and human body fluids and tissues is not submitted, since Potassium sorbate is not classified as toxic (T) or as highly toxic (T+).	
Undertaking of intended data submission [ ]		
	Evaluation by Competent Authorities	
	Use separate "evaluation boxes" to provide transparency as	
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	Potassium Sorbate	
		Page 1 of 1
Section A4.3 Annex Point IIIA 6.1	Analytical methods including recovery rates and the limits of determination of the active substance, and for residues thereof, in/on food or feedstuffs and other products where relevant.	
	JUSTIFICATION FOR NON-SUBMISSION OF DATA	Official use only
Other existing data [ ]	Technically not feasible     Scientifically unjustified   ]	
Limited exposure   X	Other justification   ]	
Detailed justification:	An analytical determination of residues in/on food or feedstuff is only required if the active substance or the material treated with it is to be used in a manner which may cause contact with food or feedstuff or intended to be placed on, in or near soils in agricultural or horticultural use. The recommendations for use of Potassium sorbate as a wood protection agent clearly precludes any such contamination, which is why the submission of such methods is not considered to be necessary.	
Undertaking of intended data submission		

Evaluation by Competent Authorities		
	Use separate "evaluation boxes" to provide transparency as to the comments and views submitted	
	EVALUATION BY RAPPORTEUR MEMBER STATE (*)	
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Evaluation of applicant's justification		
Conclusion		
Remarks		
	COMMENTS FROM	
Date		
Evaluation of applicant's justification		
Conclusion		
Remarks		